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Rapid Selective Recycling of Spent LiFePO_4 Cathodes via a Deep Eutectic Solvent-Assisted Carbothermal Shock Method

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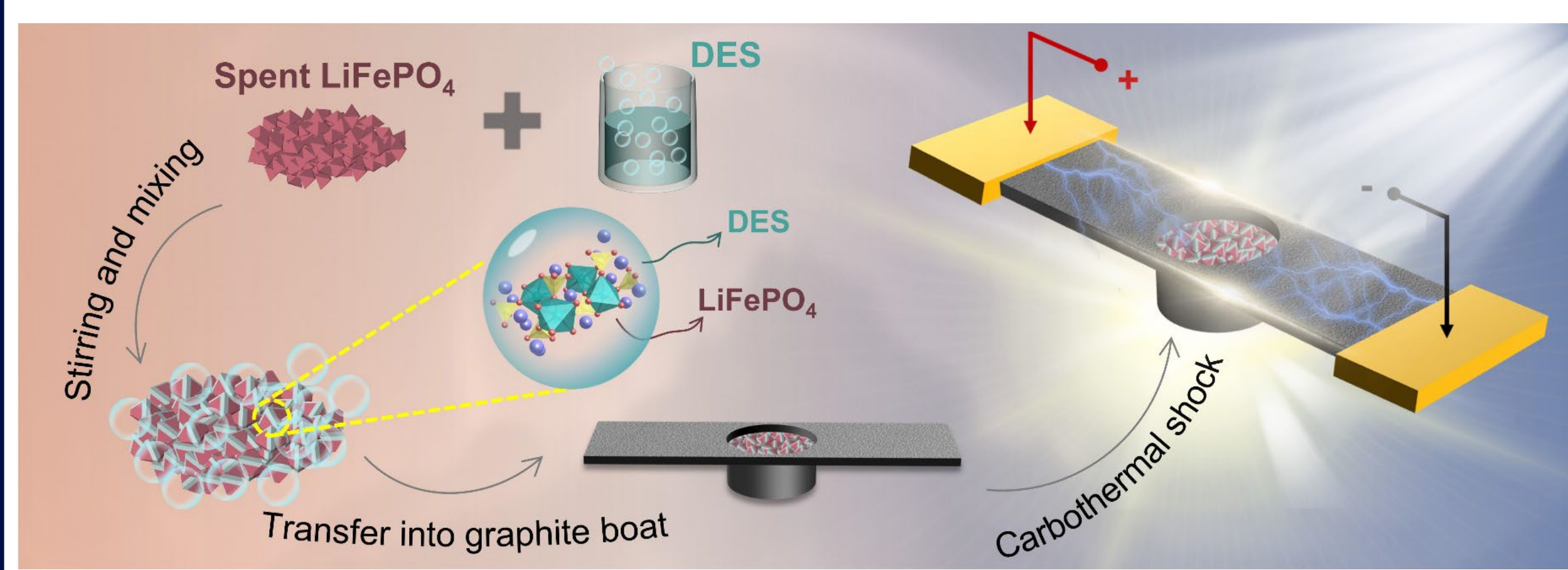
Introduction



The surge in lithium-ion battery adoption underscores the need for innovative recycling technologies, particularly for **LiFePO_4 (LFP)**, whose thermally stable olivine structure leads to high energy consumption, lithium loss, and low economic viability in conventional pyrometallurgical routes. Here, we introduce a **deep-eutectic-solvent-assisted carbothermal shock** process that achieves ultrafast (~ 20 s) and complete decomposition of spent LFP through uniform ChCl -urea encapsulation, preventing Li volatilization during the short reaction time. The method yields **high-purity Fe_2P and Li_3PO_4** with **excellent Li (97.39%) and Fe (99.17%) recovery**, enables simple magnetic separation, and provides a synergistic valorization pathway through the catalytic performance of recovered Fe_2P in PET alcoholysis.

Methods

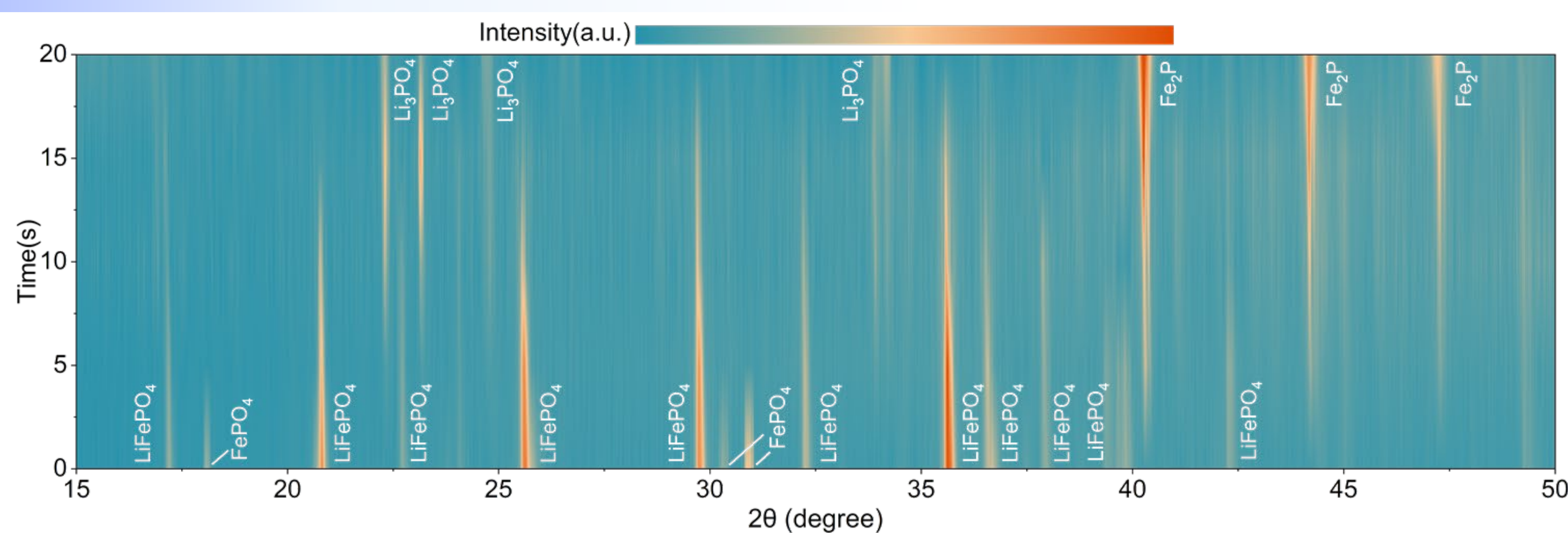
A viscous ChCl -urea deep eutectic solvent (DES) uniformly coats spent LiFePO_4 (S-LFP), enabling efficient contact and preventing Li volatilisation.



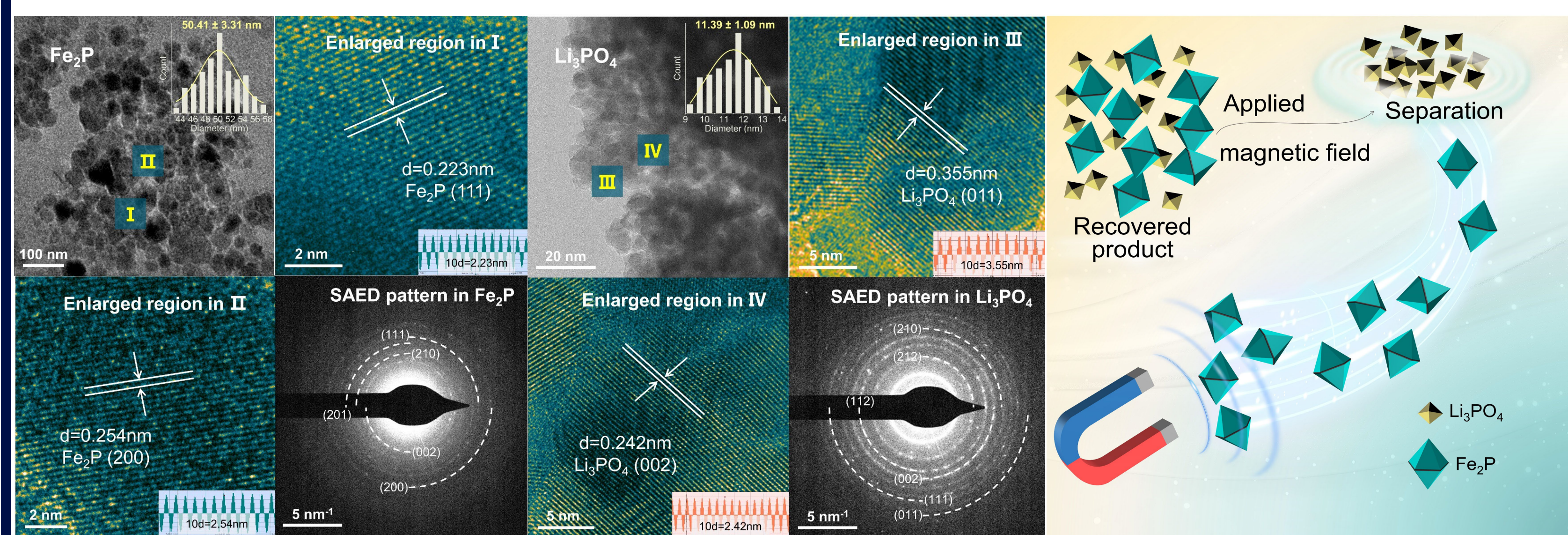
Conclusions

- A DES-assisted carbothermal shock method was developed for the efficient recycling of spent LiFePO_4 cathodes.
- This recycling method enabled complete and rapid decomposition of S-LFP into high-value Li_3PO_4 and Fe_2P .
- Li_3PO_4 and Fe_2P were easily separated through simple magnetic separation.
- Fe_2P exhibited excellent catalytic activity in PET methanolysis.
- This work offers an environmentally friendly and economically viable strategy for S-LFP recycling and provides a synergistic pathway for high-value utilization of plastic waste.

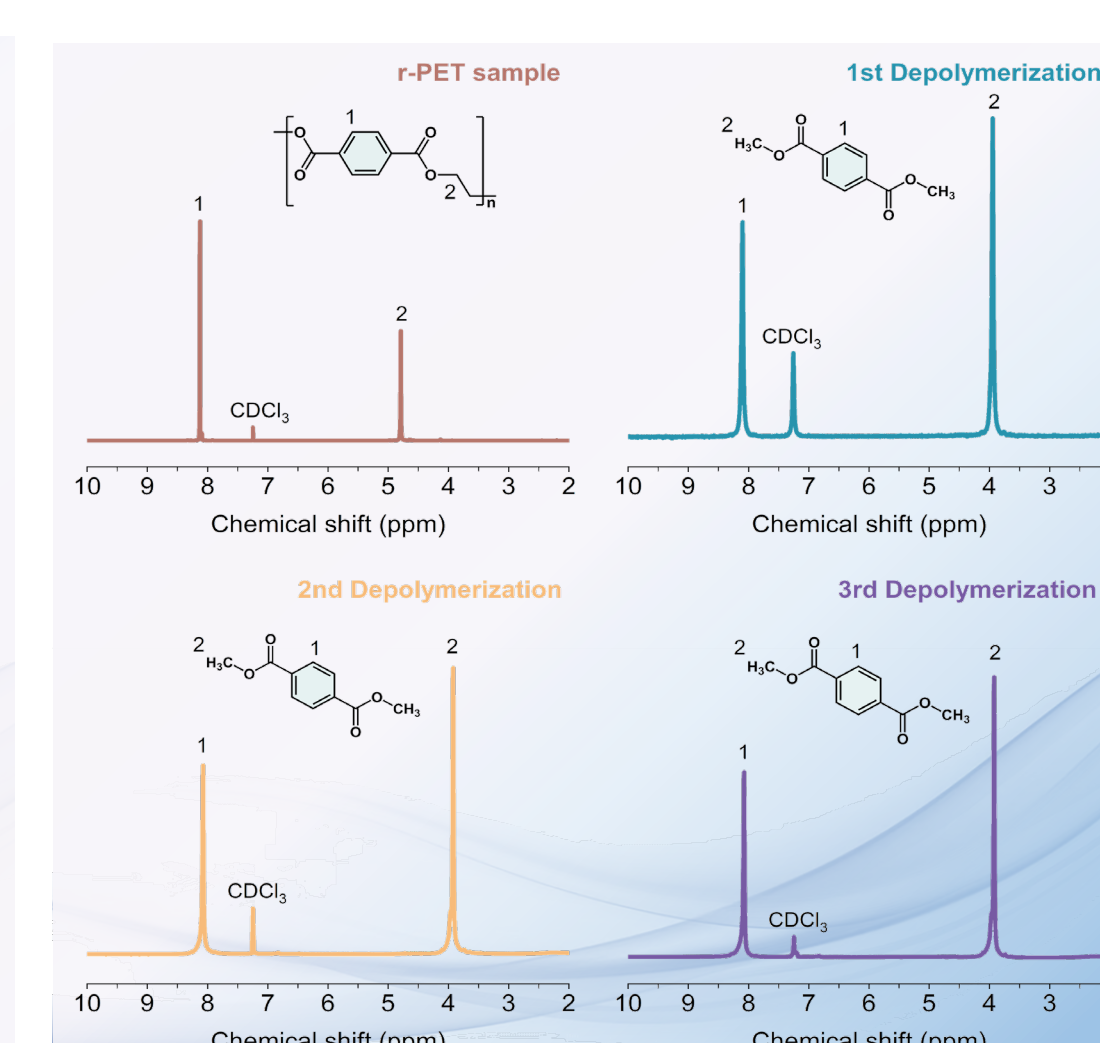
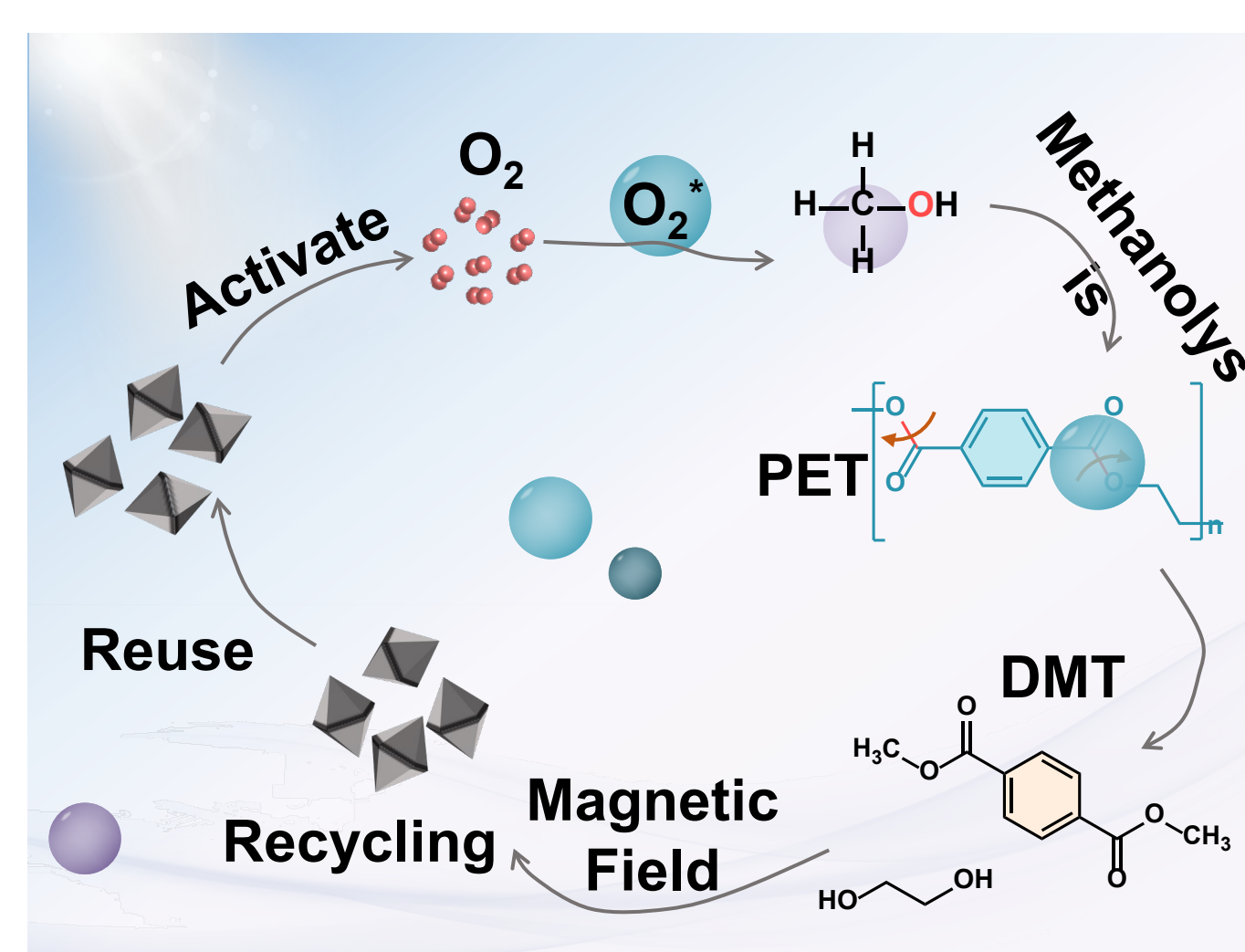
Results



Complete Transformation of S-LFP into Fe_2P and Li_3PO_4 .



TEM confirm uniform Fe_2P and Li_3PO_4 nanoparticles through efficient magnetic separation.



Fe_2P Catalyst is recyclable for reuse.

Recovered Fe_2P promotes methanolysis of Polyethylene terephthalate (PET), yielding 98% conversion and 99.5% pure dimethyl terephthalate (DMT).

Reference:

- X. Zhu, Y. Wang, et al., *Small*, 2025, e09221.
X. Zhu, Y. Wang, et al., *ACS Nano*, 2025, 19(42), 36982-36993.
X. Zhu, Y. Wang, et al., *Angewandte Chemie*, 2023, 62(15), e202300074



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