MEInetwork23



Melbourne Energy Institute

MEInetwork23 Seminar #2:

Uranium mining and refining

Speaker: Dr Neilesh Syna, Senior Hydrometallurgist, ANSTO

Moderator: Assoc Prof Kathryn Mumford, Head of Department, Chemical Engineering

8 June 2023

@MEIunimelb #MEInetwork23



MEInetwork23 Seminar Series

Seminar topic	Month
Crude oil and product supply chains - Nicholas James, VIVA Energy	Recording available online
Uranium mining and refining	8 June
Energy commodity trading	6 July
New energy commodities and critical minerals	10 August
Fiscal policy to support future energy commodity exports	7 September

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Uranium mining and milling to UOC

Neilesh Syna, ANSTO Minerals Kathryn Mumford, Melbourne Energy Institute

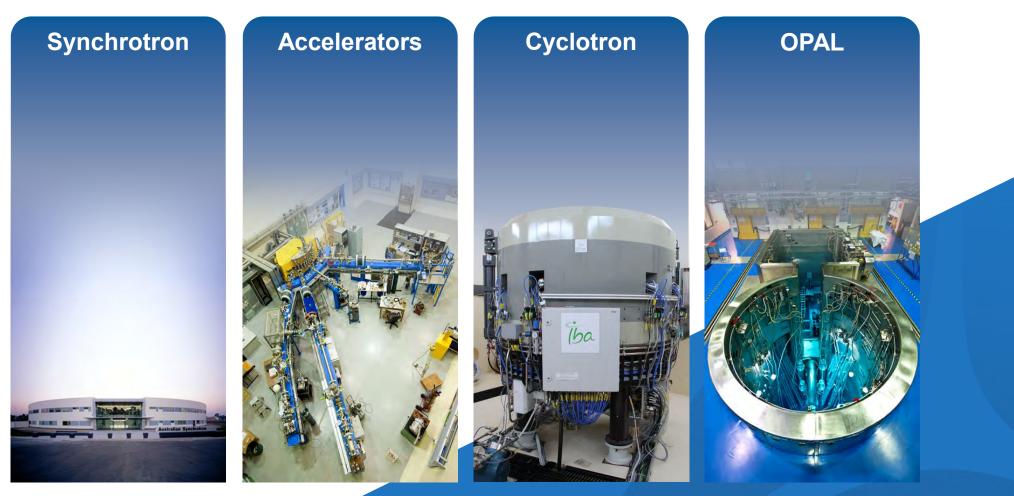
MEI*network*23 8 June 2023, Melbourne, VIC

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Science. Ingenuity. Sustainability.

Commercial Products and Services



Business Development



Detection and Imaging



Innovation and Commercialisation





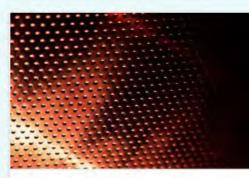
Minerals



Nuclear Waste Solutions



Radiation Services



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Continuous Mini-Pilot





Background

Fast Facts

Short Break with Questions

Different processing routes

Short Break with Questions

Recent innovations

Australia operations & projects

Future challenges & opportunities

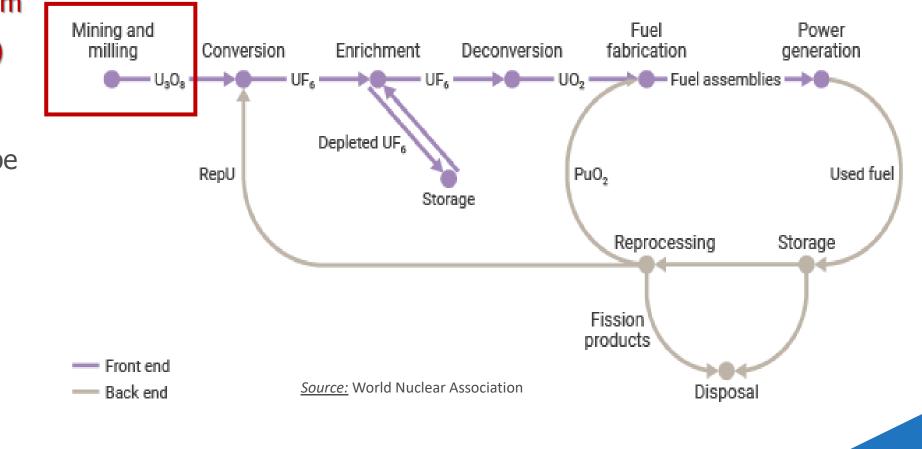


Nuclear power generation



Nuclear fuel cycle

- Exploration, Mining and
- Milling to produce Uranium Oxide Concentrate (UOC)
- Conversion to UF₆
- Enrichment of ²³⁵U isotope
- Fuel (UO₂) fabrication
- Power generation
- Spent fuel storage
- Reprocessing (U and Pu recovery)
- Waste storage and disposal



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Uranium market pricing



<u>Source:</u> BHP

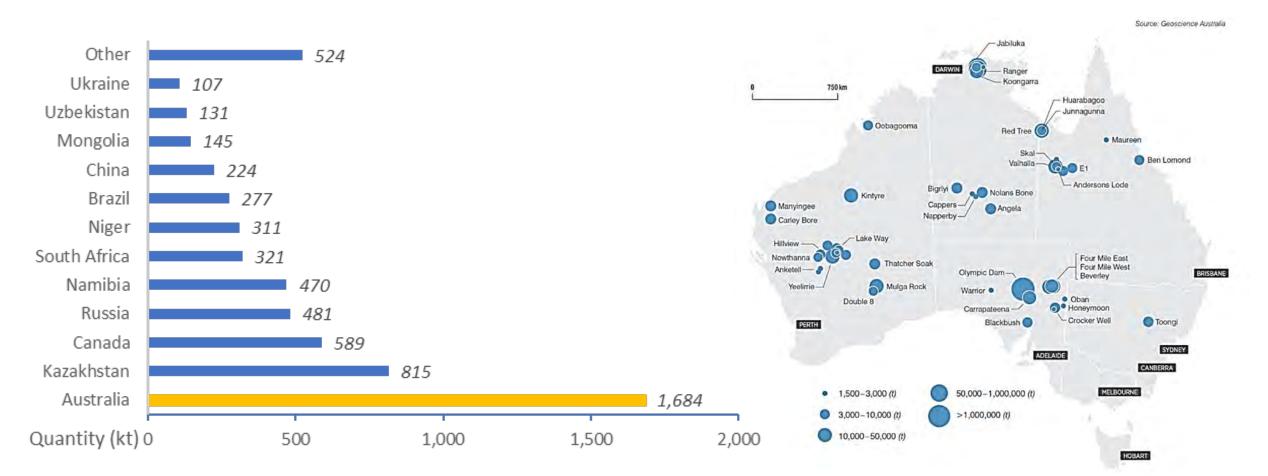
- Spot contracts
- Long-term contracts
- Financial instruments active in spot market

LT price (~US\$60-80) crucial to incentivise new mine development





Resources (2021)



Source: World Nuclear Association

ANSTO

(47 342 tU, as of 1 January 2021)

Canada

Namibia

12%

Uzbekistan*

Niger

6%

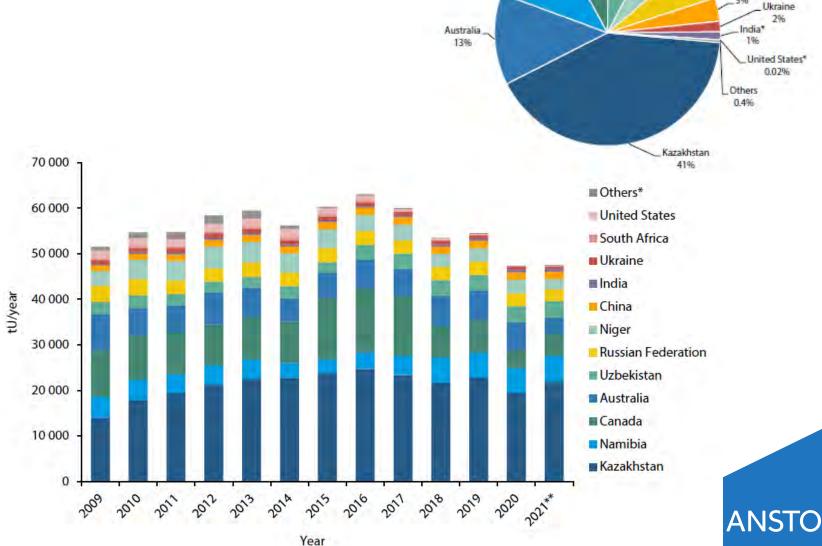
Russia 6%

> China* 3%

UOC production (2020)

- 17 countries (Australia is 2nd)
- 7 countries accounted for 94%
- From the deserts of Africa,
 through the plains of Central
 Asia to the arctic tundras of
 Canada

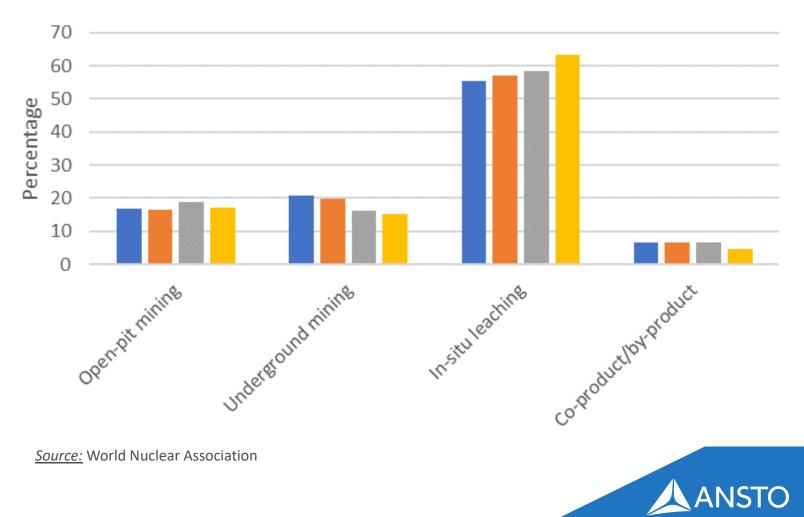
Source: OECD U Red book (2022)



Methods of production (2021)

- Open pit, underground, insitu leach and co-/byproduct
- In-situ leach is the largest and continues to grow (underground mining in decline)
- Co-/by-production typically from gold, base metals (copper and nickel) and phosphate operations

■ 2018 ■ 2019 ■ 2020 ■ 2021



Producers (2021)

- 10 companies market 94% of global mine production
- 80% of global production
 concentrated within 6 companies
- 5 are State Own Enterprises







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Break Time: Questions?



Process selection

Dependent upon:

- Resource
- Mining method
- Ore mineralogy
- U grade
- Geographical location
- Economics
- Laws and regulations
- Schedule, . . .



Mineralogy

Mineralisation:

- Vanadates
- Carbonates
- Oxides
- Phosphates
- Arsenates
- Sulphides,
- Molybdates, . . .







<u>Source:</u> Mindat

Uraninite

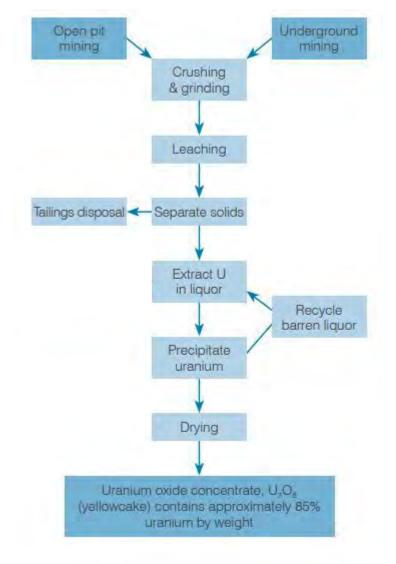
Principle Uranium minerals:

- Pitchblende
- Uraninite
- Carnotite
- Coffinite
- Brannerite
- Torbenite
- Wulfenite, . . .

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Conventional process

- 1. Mining
- 2. Crushing & grinding
- 3. Leach
- 4. Solid-liquid (S/L) separation
- 5. Purification & enrichment
- 6. Precipitation & dewatering
- 7. Drying & calcination
- 8. Packaging & transport
- 9. Waste treatment



Source: World Nuclear Association







Open-pit mining

Source: Energy Resources Australia

- Near surface ores (< 100 m)</p>
- Requires heavy equipment for soil and waste rock removal before ore access
- Pros:
 - $\,\,$ > Better ventilation, less costly than UG
- Cons:
 - > Huge footprint, expensive remediation

Underground mining

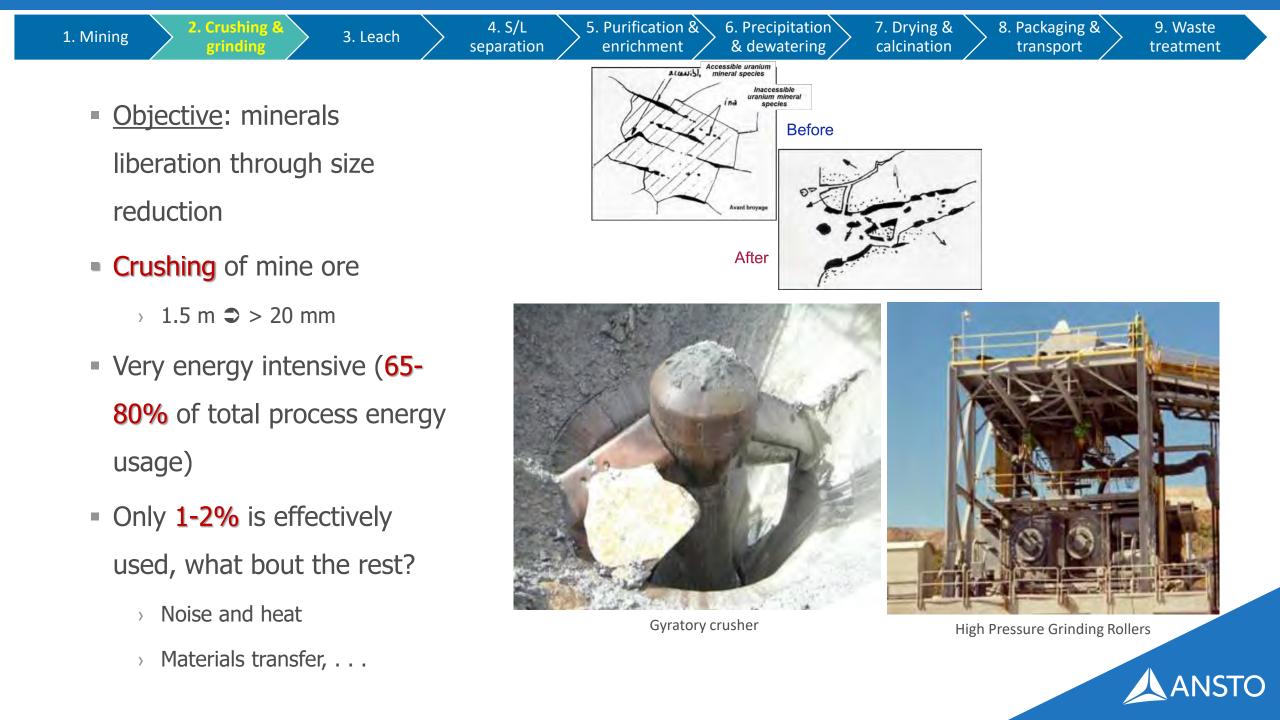
- Deep orebodies (> 100 m)
- Requires vertical shafts to ore depth followed by creation of tunnels, ramps and chambers

McArthur River (Canada)

- Pros:
 - > Smaller footprint, less waste
- Cons:
 - > Expensive operation and remediation



Source: Cameco Corp





- Crushed ore storage prior to beneficiation and/or grinding circuits
- Options of open or covered storage
- Ensures undisrupted feed to downstream process units





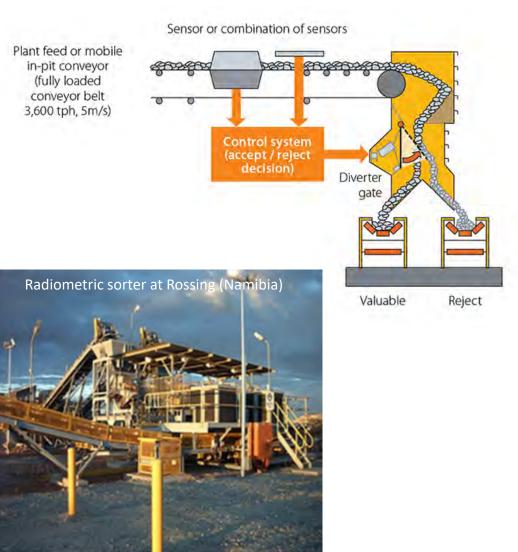
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<u>Source:</u> Orano



- **Objective:** Increase U grade whilst reducing tonnage treated
- Available techniques include:
 - Sizing >

- Gravity spirals
- Magnetic & electrostatic
- Flotation
- Sorting >
- Very few U ores are amenable to the above techniques except for (radiometric) sorting
 - Rossing (Namibia)



ANSTO

Source: Applied Sorting Technologies

4. S/L 5. Purification & separation enrichment

6. Precipitation & dewatering

7. Drying & 8. Packaging & calcination

transport

9. Waste treatment

2nd stage grinding circuit:

2. Crushing &

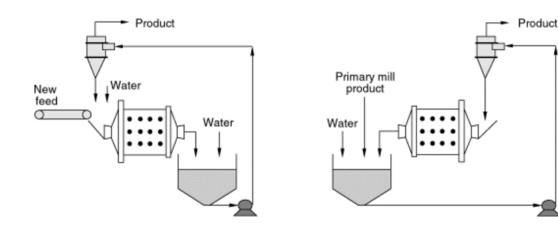
grinding

1. Mining

(Autogenous/Semi-autogenous) mills, $> 100 \mu m$ >

3. Leach

- (Ball/Rod) mills, $> 30 \mu m$ >
- (Stirred/Tower/ISA) mills, >1 µm >
- Operated either wet or dry
- Use grinding media (metallic or ceramic balls)
- Typically operated in closed loop arrangement





Source: Energy Resources Australia



Source: Orano



5. Purification & 6. Precipitation enrichment & dewatering

7. Drying & 8. Packaging & calcination

transport

9. Waste treatment

Objective: Chemical dissolution (leach) of U minerals & limiting gangue dissolution

4. S/L

separation

- Reagent choice of:
 - Acid (most common)

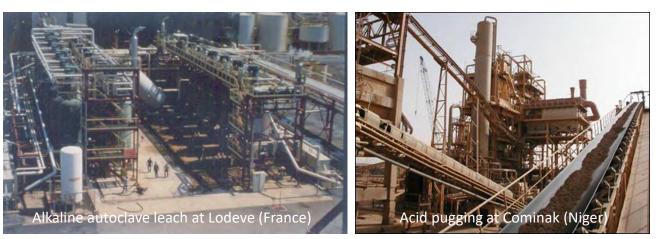
2. Crushing &

grinding

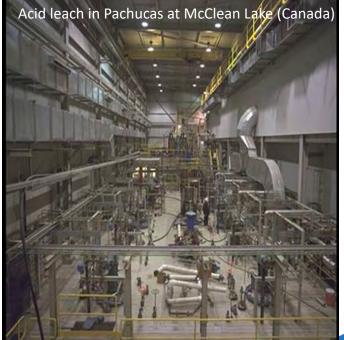
- Alkaline (for high acid consuming ores, e.g. carbonates)
- Variables include:
 - > Mineralogy, grade, time, temperature, . . .
- Options:

1. Mining

- Dynamic leach (vat, Pachuca, agitated tank, autoclave)
- Heap leach
- In-situ leach



Source: Orano



Source: Orano



1. Mining

3. Leach

4. S/L separation

5. Purification & 6. Precipitation enrichment & dewatering

7. Drying & calcination

9. Waste treatment

- Objective: Separation of U bearing liquor from leach solids for further processing
- Variables include:

2. Crushing &

grinding

- > Leach method, clay content, particle size, water availability, solution grade, costs, . . .
- Options:
 - Thickening followed by filtration (belt, drum, pressure, . . .)
 - Counter-current decantation (CCD)
 - Resin-in-Pulp (RIP)



Source: Orano



8. Packaging &

Source: Paladin Energy





1. Mining

> 3. Leach

4. S/L separation

5. Purification 6. Precipitation & dewatering

7. Drying & calcination 9. Waste treatment

<u>Objective:</u> Purification &
 concentration of U from contaminants

2. Crushing &

grinding

- Variables include:
 - Solution flow, grade, pH,
 impurities, . . .
- Options:
 - > Solvent extraction (SX)
 - » Mixer settlers, pulsed columns
 - > Ion exchange (IX) resins
 - > Resin-in-Pulp (RIP)



Source: Energy Fuels Inc.



<u>Source:</u> Dow

<u>Source:</u> BHP

8. Packaging &

transport



Source: Heathgate Resources



6. Precipitation

& dewatering

9. Waste treatment

Resin-in-pulp (RIP)

2. Crushing &

grinding

1. Mining

Negates the need for S/L step before contact

3. Leach

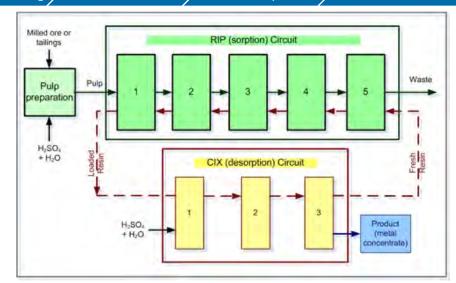
4. S/L

separation

5. Purification

& enrichment

- Process pulps with 50wt% solids
- Coarser resin beads retained in baskets
- Consists of 2 (adsorption and desorption) circuits
- Higher CAPEX but lower OPEX (over standard IX)
- Issues with resin fracture and U loss





Source: Lotus Resources



7. Drying & calcination 8. Packaging & transport

9. Waste treatment

 <u>Objective</u>: Transform U from solution to solid phase by increasing solution pH

3. Leach

4. S/L

separation

5. Purification &

enrichment

6. Precipitation

& dewatering

Variables include:

1. Mining

- Solution flow, environmental regulations, reagent availability, temperature, converter requirements,
- Reagent options include:

2. Crushing &

grinding

- > Ammonia, peroxide, caustic, lime or milk of magnesia
- Tank precipitation (either in batches or continuously) followed by filtration
- Step can be repeated to further reject more impurities (Mg, Na, Zr, etc.)
- Converters prefer UO₄ or U₃O₈ concentrates





Source: Orano



5. Purification & 6. Precipitation enrichment & dewatering

ROX. 10 m

7. Drying & calcination

8. Packaging & 9. V transport trea

9. Waste treatment

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 <u>Drying objective</u>: Displace water from wet cake and crystalline water

3. Leach

4. S/L

separation

Variables include:

2. Crushing &

grinding

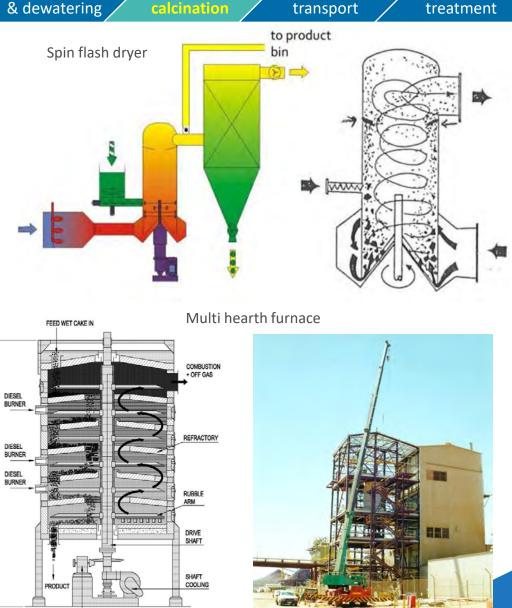
1. Mining

- > Precipitate type, temperature, . . .

 $UO_4.2H_2O \leftrightarrow UO_4 + 2H_2O$

- Calcination objective: Convert to final product
- Variables include:
 - > Product type, temperature, . . .
 - » e.g. ammonia precipitate (ADU) at 650 800°C

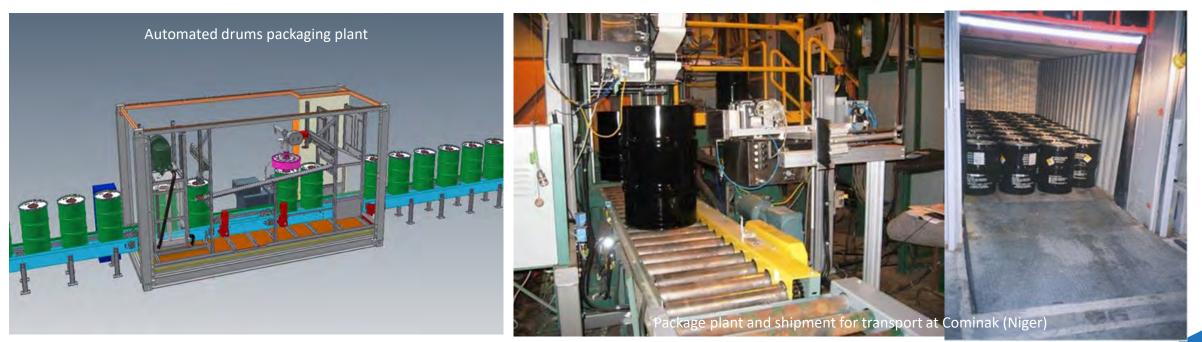
 $9(NH_4)_2U_2O_7 \leftrightarrow 6U_3O_8 + 14NH_3 + 15H_2O + 2N_2$



DRIVE



- Objective: Package end product for delivery to converters (chosen by customers)
- Automated to minimise radioactive contamination and exposure



Source: Adelaide Control Equipment

<u>Source:</u> Orano



1. Mining

grinding

4. S/L separation

5. Purification & 6. Precipitation enrichment & dewatering

7. Drying & calcination

- Uranium oxide (U_3O_8) is sold via both contract and spot market
- Each sale is controlled through the U3O8 specification
 - Some converters have their own specification
 - An increasing number use ASTM C967 (for naturally occurring uranium, minimum 65 % U or 85 % U3O8)

Limits:

- Without penalty: impurities up to indicated level attract no cost penalty
- Without rejection: above this the consignment can/will be rejected
- In between: Sliding scale of cost, based upon element and concentration, subtracted from the unit price.
- Why U-234 limit?
 - Remember natural abundance U-234 is 0.0055 % (55 ppm)
 - U-234 absorbs neutrons and affects the breeding properties of the fuel
 - Shortens life of fuel rods
 - Shorter half life means heightened exposure control
 - Compensate by increasing U-235 content in fuel (issues, plus cost)

C967 - 13

8. Packaging 8

transport

Impurity	Maximum Concentration Limit (Mass %, Uranium Basis) ⁴		
	Limit Without Penalty ^B	Limit Without Rejection	
As	0.05	0.10	
В	0.005	0.10	
Ca	0.05	1.00	
Carbonate	0.20	0.50	
F	0.01	0.10	
Halogens ^D	0.05	0.10	
Fe	0.15	1.00	
Mg	0.02	0.50	
Moisture	2.0	5.0	
Mo	0.10	0.30	
P	0.10	0.70	
к	0.20	3.00	
Si(calculated as SiO ₂)	0.50	2.50	
Na	0.50	7.50	
S	1.00	4.00	
Th	0.10	2.50	
TI	0.01	0.05	
V	0.06	0.30	
Zr	0.01	0.10	
234U	56 ^F	62 ^F	

Source: ASTM



1. Mining

4. S/L separation



7. Drying & 8. Packaging & calcination

transport

9. Waste treatment

Objectives:

Disposal of **solid tailings** compliant to local laws

2. Crushing &

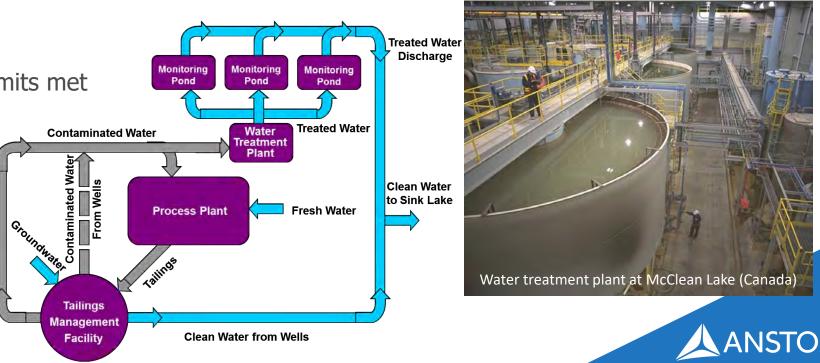
grinding

Disposal of liquid effluents by immobilising environmental contaminants (heavy metals and radioactive) and ensuring local limits met for water discharge

» e.g.
$$Fe^{3+}$$
 + $3OH^-$ → $Fe(OH)_3$ ^{Mater} Mater Mate

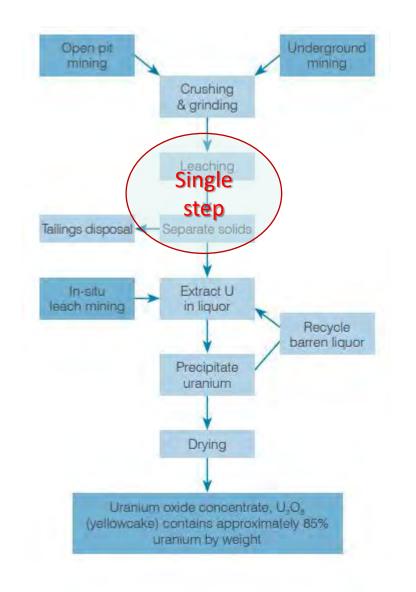


Source: Orano



Heap leach process

- 1. Mining
- 2. Crushing & agglomeration
- 3. Heap leach
- 4. Purification & enrichment
- 5. Precipitation & dewatering
- 6. Drying & calcination
- 7. Packaging & transport
- 8. Waste treatment

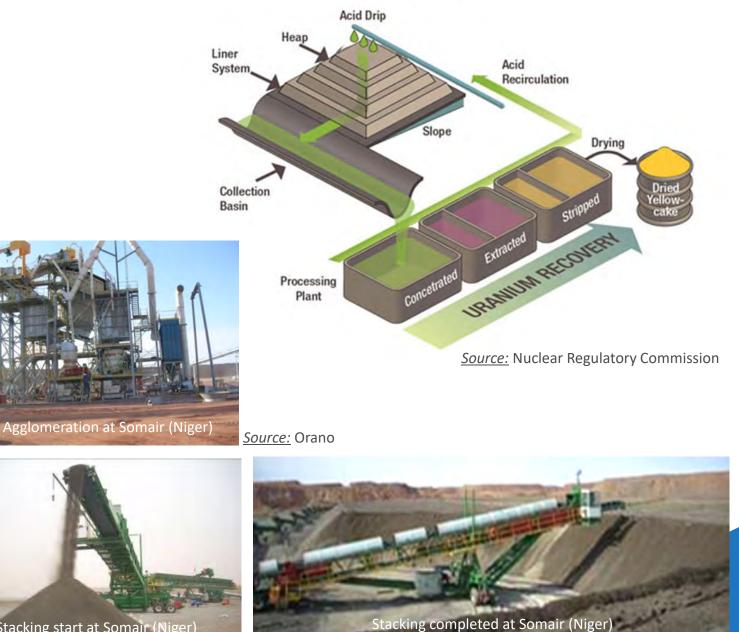




Heap leach process

- Suitable for low grade ores
- Ores crushed and agglomerated with reagents then built into heaps and irrigated with leach solution
- Leach solution recovered from collection basin
- Slow reaction rate (3 9 m)
- Moderate CAPEX with low OPEX
- SX or IX used for U recovery
- Used in Niger & Finland

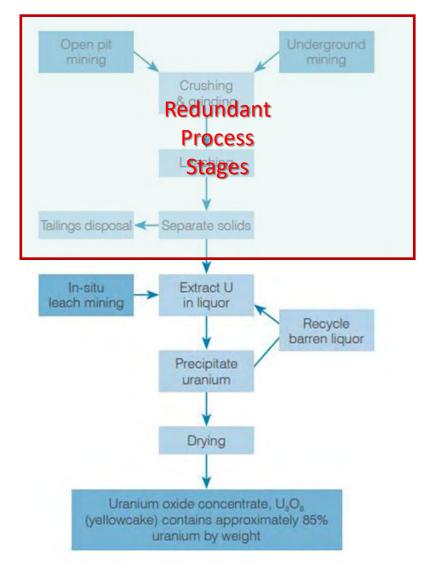
The Heap Leach Recovery Process



Stacking start at Somair (Niger

In-situ leach process

- 1. **ISL** involves drilling, well preparation & cell irrigation
- 2. Purification & enrichment
- 3. Precipitation & dewatering
- 4. Drying & calcination
- 5. Packaging & transport
- 6. Waste treatment



Source: World Nuclear Association



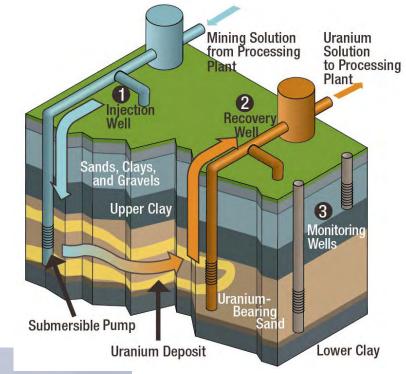
In-situ leach process

- Suitable for ore bodies
 sandwiched between clay layers)
- Reagents injected into ore body (18-24m)
- Leach solution is pumped out for
 U recovery
- Monitoring wells check for chemical escapes
- Low CAPEX and low OPEX
- Little waste generation & minimal surface ground disturbance
- IX used for U recovery



Production well at Tortkuduk (Kazakhstan)





Source: Nuclear Regulatory Commission

Well field at Smith Ranch-Highland (USA)

Source: Cameco Corp.

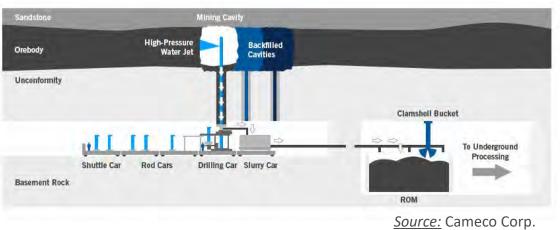


Break Time: Questions ?



Recent innovations in mining

Jet boring



- UG mining technique used at Cigar (Canada)
- Freezing the ore and surrounding rocks using chilled brine (-40°C) in large pipes
- Production tunnel created for jet boring system entry
- Pilot hole drilled through orebody then jet boring nozzle inserted
- High pressure water jet extracts ores
- Places workers UG

SABRE



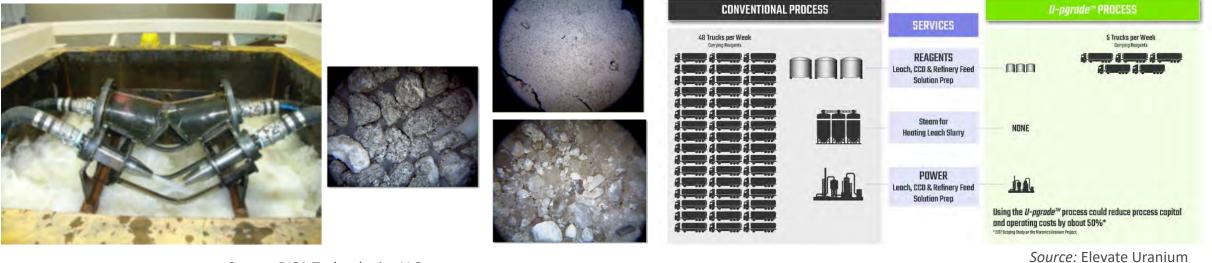
<u>Source:</u> Orano

- Surface Access Borehole Resource Extraction
- Non-entry surface based mining technique
- Benefits of:
 - > Economics (flexibility and scalability)
 - > Safety (health and injury minimised)
 - > Environmental (small footprint and lower water and energy demand)
- Potential for Midwest project (Canada)



Recent innovations in processing

Abrasion based beneficiation technique



Source: DISA Technologies LLC

- 2 proponents
 - U-pgrade[™] (Elevate Uranium) and High Pressure Slurry Ablation technique, HPSA (DISA Technologies LLC)
- Uses the forces of abrasion, elastic compression and rebounding
- Benefits of:
 - Lower mine haulage
 - Lower milling costs (throughput, reagents, energy)
 - Better environmental outcomes (lower (mill) tailings and cleaner (mine) waste dumps)
- Tested using Colorado (USA) sandstones and Namibian calcretes



Recent innovations in processing Nanofiltration Saline IX (ANSTO)



Source: BMS Engineers

- Filtration technique employing membranes under high pressure
- Applied for reagent recovery
- Used at Langer Heinrich (Namibia) & Kayelekera (Malawi)



Source: Boss Resources

NSTO

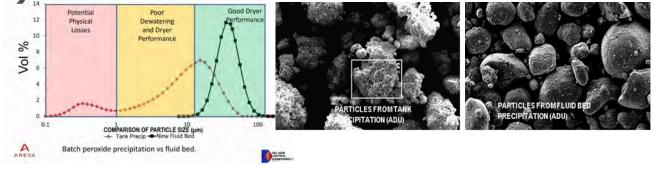
- Ideal for locations with saline (Cl) water
- High capacity Strong Base Anion (SBA) resin binds strongly to U
- Patented U elution method developed
- To be applied at the restart of Honeymoon (SA)

Recent innovations in processing

Fluidised Bed Precipitation (Orano)



Particle Size Distribution





- Larger particles

 better S-L separation
- Less fines **C** reduced dust generation
- Higher bulk density
 transport savings
- Used at KATCO JV (Kazakhstan) and Somair (Niger) operations

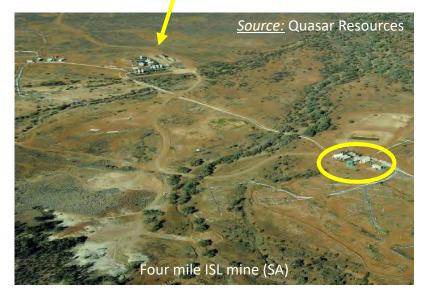


<u>Source:</u> Orano

Source: BHP

Australian operations (2021)

Beverly ISL mine (SA)

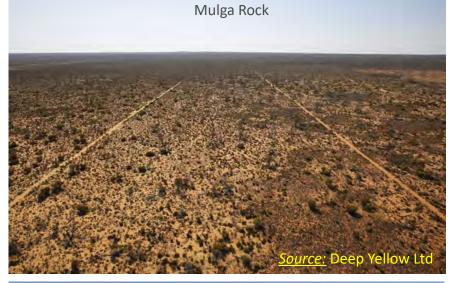








Australian projects (2021)



<u>Source:</u> Toro Energ





Source: Geoscience Australia





Future challenges & opportunities

- Process & economics:
 - » Treat lower grade ores and to reprocess old tailings
 - » Produce U as a by-product (e.g. Au, Cu, REE, . . .) and/or produce by-products from U deposits (e.g. Co, Ni, V, Sc, Re, . . .)
 - » Adopt and integrate alternative process technologies (e.g. bioleach, . . .)
- Safety: Provide safe work conditions & reduce worker exposure,
- Environment: Reduce impacts & consider sustainable development.
- Laws and Regulations: Adapt to changing landscape
- Social licence: Transparent engagement with local population and the wider public



Thank you. Questions?







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